

# 1,2,3,4-Tetrahydrophenazine 5,10-dioxide

Tao Sun,<sup>a</sup> Jianye Li,<sup>a</sup> Hongwei Qiao,<sup>b</sup> Aiyou Hao<sup>a\*</sup> and  
Yueming Li<sup>a</sup>

<sup>a</sup>School of Chemistry and Chemical Engineering and Key Laboratory of Colloid and Interface Chemistry of the Ministry of Education, Shandong University, Shanda Nanlu 27, Jinan 250100, People's Republic of China, and <sup>b</sup>Shandong Shengquan Chemical Co. Ltd, Zhangqiu Jinan, 250204, People's Republic of China

Correspondence e-mail: haoay@sdu.edu.cn

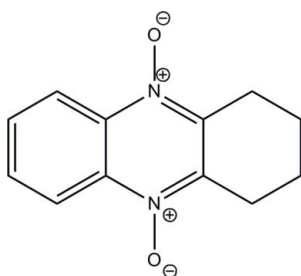
Received 10 June 2010; accepted 29 July 2010

Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å; disorder in main residue;  $R$  factor = 0.040;  $wR$  factor = 0.127; data-to-parameter ratio = 10.2.

The complete molecule of the title compound,  $\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}_2$ , lies on two crystallographic symmetry elements: a twofold axis and a mirror plane. In the molecular structure, the quinoxaline ring and two methylene substituents lie on the mirror plane while the other two methylene groups are disordered about the plane. The crystal packing is stabilized by weak intermolecular  $\pi-\pi$  stacking interactions with centroid-centroid distances of 3.6803 (7) Å.

## Related literature

For the synthetic preparation, see: Haddadin & Issidorides (1965); Issidorides & Haddadin (1966). For background to quinoxaline di-*N*-oxide compounds, see: Edwards *et al.* (1975) and for their biological activity, see: Urquiola *et al.* (2008). For a related structure, see: Wang *et al.* (2010).



## Experimental

### Crystal data

$\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}_2$	$V = 1018.80$ (3) Å <sup>3</sup>
$M_r = 216.24$	$Z = 4$
Orthorhombic, <i>Cmcm</i>	Mo $K\alpha$ radiation
$a = 11.7780$ (2) Å	$\mu = 0.10$ mm <sup>-1</sup>
$b = 13.1938$ (3) Å	$T = 296$ K
$c = 6.5561$ (1) Å	$0.31 \times 0.29 \times 0.26$ mm

### Data collection

Bruker APEXII CCD area-detector diffractometer	3311 measured reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	620 independent reflections
$T_{\min} = 0.67$ , $T_{\max} = 0.74$	534 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.016$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.127$	$\Delta\rho_{\text{max}} = 0.23$ e Å <sup>-3</sup>
$S = 1.10$	$\Delta\rho_{\text{min}} = -0.29$ e Å <sup>-3</sup>
620 reflections	
61 parameters	

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: SHELXTL (Sheldrick, 2008); molecular graphics: SHELXTL; software used to prepare material for publication: WinGX (Farrugia, 1999).

This work was supported by the NSFC (grant No. 20625307), the National Basic Research Program of China (973 Program, 2009CB930103) and the Graduate Independent Innovation Foundation of Shandong University (GIIFSDU).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NK2042).

## References

- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
- Bruker (2007). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Edwards, M. L., Bambury, R. E. & Ritter, H. W. (1975). *J. Med. Chem.* **18**, 637–639.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Haddadin, M. J. & Issidorides, C. H. (1965). *Tetrahedron Lett.* **6**, 3253–3256.
- Issidorides, C. H. & Haddadin, M. J. (1966). *J. Org. Chem.* **31**, 4067–4068.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Urquiola, C., Cabrera, M., Lavaggi, M. L., Cerecetto, H., Gonzalez, M., Cerain, A. L., Monge, A., Costa-Filho, A. J. & Torre, M. H. (2008). *J. Inorg. Biochem.* **102**, 119–126.
- Wang, Z., Jia, W., Yao, H., Qiu, H. & Wang, W. (2010). *Acta Cryst.* **E66**, o1380.

**supplementary materials**

*Acta Cryst.* (2010). E66, o2425 [ doi:10.1107/S1600536810030242 ]

## 1,2,3,4-Tetrahydrophenazine 5,10-dioxide

T. Sun, J. Li, H. Qiao, A. Hao and Y. Li

### Comment

Quinoxaline di-N-oxide compounds are widely used in sterilization and growth-promoting of animals, pharmacological properties usable as intermediates for producing plant protection agents (Edwards *et al.*, 1975). There has been a growing interest in the syntheses of quinoxaline di-N-oxide compounds that have both biological and commercial importance (Urquiola *et al.*, 2008). Now, we report herein the crystal structure of the title benzotriazole derivative.

The complete molecule of the title compound, C<sub>12</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>, is generated by a crystallographic symmetry operation along a twofold axis. In the molecular structure of the crystal, the quinoxaline ring and two methylene substituents of the quinoxaline ring locate at a mirror plane of the *Cmcm* group. The other two methylenes of the cyclohexane ring are disordered over two positions with half occupancy. The crystal packing is stabilized by weak intermolecular  $\pi$ - $\pi$  aromatic stacking interactions with centroid-centroid distances of 3.6803 (7) Å.

### Experimental

The compound was synthesized as described previously by Haddadin & Issidorides (1965) and Issidorides & Haddadin (1966). Yellow crystals were obtained by slow evaporation of a methanolic solution.

### Refinement

H atoms in the benzene were placed in geometrically calculated positions and refined using a riding model. H atoms in CH<sub>2</sub> groups were located in geometrically calculated positions also but their positions were refined independently and their isotropic displacement parameters were fixed to 0.08 in the refinement. Two CH<sub>2</sub> groups were disordered over symmetry elements and refined with half occupancy.

### Figures

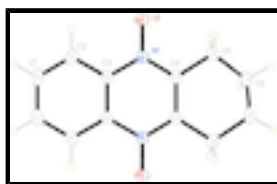


Fig. 1. **Figure 1.** A view of the title compound with displacement ellipsoids are drawn at the 30% probability level. Unlabelled atoms are related to labelled atoms by a twofold rotation. The second disorder component is omitted.

## 1,2,3,4-Tetrahydrophenazine 5,10-dioxide

### Crystal data

C<sub>12</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>

$F(000) = 456$

# supplementary materials

---

$M_r = 216.24$	$D_x = 1.410 \text{ Mg m}^{-3}$
Orthorhombic, $Cmcm$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: $-C 2c 2$	Cell parameters from 1577 reflections
$a = 11.7780 (2) \text{ \AA}$	$\theta = 2.3\text{--}26.8^\circ$
$b = 13.1938 (3) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$c = 6.5561 (1) \text{ \AA}$	$T = 296 \text{ K}$
$V = 1018.80 (3) \text{ \AA}^3$	Prism, yellow
$Z = 4$	$0.31 \times 0.29 \times 0.26 \text{ mm}$

## Data collection

Bruker APEXII CCD area-detector diffractometer	620 independent reflections
Radiation source: fine-focus sealed tube graphite	534 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans	$R_{\text{int}} = 0.016$
Absorption correction: multi-scan ( <i>SADABS</i> ; Sheldrick, 1996)	$\theta_{\text{max}} = 26.9^\circ$ , $\theta_{\text{min}} = 2.3^\circ$
$T_{\text{min}} = 0.67$ , $T_{\text{max}} = 0.74$	$h = -14 \rightarrow 14$
3311 measured reflections	$k = -16 \rightarrow 12$
	$l = -8 \rightarrow 8$

## Refinement

Refinement on $F^2$	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.040$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.127$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.10$	$w = 1/[\sigma^2(F_o^2) + (0.0854P)^2 + 0.1004P]$
620 reflections	where $P = (F_o^2 + 2F_c^2)/3$
61 parameters	$(\Delta/\sigma)_{\text{max}} < 0.001$
0 restraints	$\Delta\rho_{\text{max}} = 0.23 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.29 \text{ e \AA}^{-3}$

## Special details

**Experimental.**  $^1\text{H NMR}$  (400 MHz,  $\text{DMSO-d}_6$ ):  $\delta$  8.67 (2H, d,  $J = 3.5 \text{ Hz}$ , Ar—H), 7.89 (2H, d,  $J = 3.2 \text{ Hz}$ , Ar—H), 3.77 (1H, s, CH), 2.66 (3H, s, CH<sub>3</sub>), 2.51 (2H, m, CH<sub>2</sub>), 1.45 (6H, s, CH<sub>3</sub>); Calcd for  $\text{C}_{13}\text{H}_{16}\text{N}_2\text{O}_2$ : C, 67.22; H, 6.94; N, 12.06. Found: C, 67.18; H, 6.99; N, 11.95; ESIMS calcd for  $\text{C}_{13}\text{H}_{16}\text{N}_2\text{O}_2\text{H}^+$   $m/z$  232.38, found  $m/z$  232.19.

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -

factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	0.44078 (13)	-0.14330 (10)	0.2500	0.0434 (4)	
H1	0.4017	-0.2045	0.2500	0.052*	
C2	0.38142 (12)	-0.05388 (9)	0.2500	0.0399 (4)	
H2	0.3025	-0.0544	0.2500	0.048*	
C3	0.44062 (11)	0.03818 (9)	0.2500	0.0311 (4)	
C4	0.44068 (10)	0.21722 (9)	0.2500	0.0326 (4)	
C5	0.37292 (12)	0.31337 (10)	0.2500	0.0470 (4)	
H5	0.3247 (10)	0.3113 (9)	0.131 (2)	0.070*	
C6	0.44666 (19)	0.40485 (17)	0.1867 (4)	0.0618 (9)	0.50
H6	0.407 (2)	0.4669 (19)	0.199 (5)	0.090*	0.50
H7	0.468 (3)	0.3993 (19)	0.040 (4)	0.090*	0.50
N1	0.38161 (10)	0.12976 (7)	0.2500	0.0336 (4)	
O1	0.27161 (9)	0.12938 (6)	0.2500	0.0508 (4)	

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0608 (9)	0.0288 (7)	0.0407 (7)	-0.0082 (5)	0.000	0.000
C2	0.0426 (8)	0.0339 (7)	0.0432 (7)	-0.0078 (5)	0.000	0.000
C3	0.0339 (8)	0.0281 (7)	0.0314 (6)	-0.0008 (4)	0.000	0.000
C4	0.0318 (7)	0.0277 (7)	0.0384 (7)	0.0008 (4)	0.000	0.000
C5	0.0372 (8)	0.0320 (8)	0.0719 (10)	0.0056 (5)	0.000	0.000
C6	0.0516 (11)	0.0289 (10)	0.105 (3)	0.0038 (7)	0.0002 (10)	0.0110 (10)
N1	0.0282 (6)	0.0313 (6)	0.0412 (6)	-0.0005 (3)	0.000	0.000
O1	0.0269 (6)	0.0447 (7)	0.0807 (8)	-0.0011 (3)	0.000	0.000

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

C1—C2	1.3714 (19)	C5—C6 <sup>ii</sup>	1.544 (3)
C1—C1 <sup>i</sup>	1.395 (3)	C5—C6	1.544 (3)
C1—H1	0.9300	C5—H5	0.965 (13)
C2—C3	1.4005 (17)	C6—C6 <sup>ii</sup>	0.831 (5)
C2—H2	0.9300	C6—C6 <sup>iii</sup>	1.256 (5)
C3—N1	1.3939 (15)	C6—C6 <sup>i</sup>	1.506 (4)
C3—C3 <sup>i</sup>	1.399 (2)	C6—H6	0.95 (2)
C4—N1	1.3474 (15)	C6—H7	1.00 (3)
C4—C4 <sup>i</sup>	1.397 (2)	N1—O1	1.2956 (16)
C4—C5	1.4987 (16)		
C2—C1—C1 <sup>i</sup>	120.65 (9)	C6 <sup>ii</sup> —C6—C6 <sup>iii</sup>	90.000 (2)
C2—C1—H1	119.7	C6 <sup>ii</sup> —C6—C6 <sup>i</sup>	56.5 (2)

## supplementary materials

---

C1 <sup>i</sup> —C1—H1	119.7	C6 <sup>iii</sup> —C6—C6 <sup>i</sup>	33.5 (2)
C1—C2—C3	119.49 (15)	C6 <sup>ii</sup> —C6—C5	74.39 (10)
C1—C2—H2	120.3	C6 <sup>iii</sup> —C6—C5	124.23 (10)
C3—C2—H2	120.3	C6 <sup>i</sup> —C6—C5	108.72 (16)
N1—C3—C3 <sup>i</sup>	119.90 (7)	C6 <sup>ii</sup> —C6—H6	85.1 (18)
N1—C3—C2	120.24 (14)	C6 <sup>iii</sup> —C6—H6	119.6 (16)
C3 <sup>i</sup> —C3—C2	119.86 (8)	C6 <sup>i</sup> —C6—H6	111.4 (17)
N1—C4—C4 <sup>i</sup>	121.08 (7)	C5—C6—H6	112.1 (16)
N1—C4—C5	116.74 (12)	C6 <sup>ii</sup> —C6—H7	164.4 (18)
C4 <sup>i</sup> —C4—C5	122.17 (7)	C6 <sup>iii</sup> —C6—H7	75.0 (18)
C4—C5—C6 <sup>ii</sup>	111.23 (14)	C6 <sup>i</sup> —C6—H7	108.4 (18)
C4—C5—C6	111.23 (14)	C5—C6—H7	110.3 (16)
C6 <sup>ii</sup> —C5—C6	31.2 (2)	H6—C6—H7	106 (2)
C4—C5—H5	106.8 (7)	O1—N1—C4	121.31 (9)
C6 <sup>ii</sup> —C5—H5	124.8 (8)	O1—N1—C3	119.68 (9)
C6—C5—H5	97.8 (7)	C4—N1—C3	119.01 (13)
C1 <sup>i</sup> —C1—C2—C3	0.0	C4—C5—C6—C6 <sup>i</sup>	-50.6 (2)
C1—C2—C3—N1	180.0	C6 <sup>ii</sup> —C5—C6—C6 <sup>i</sup>	45.6 (2)
C1—C2—C3—C3 <sup>i</sup>	0.0	C4 <sup>i</sup> —C4—N1—O1	180.0
N1—C4—C5—C6 <sup>ii</sup>	163.23 (11)	C5—C4—N1—O1	0.0
C4 <sup>i</sup> —C4—C5—C6 <sup>ii</sup>	-16.77 (11)	C4 <sup>i</sup> —C4—N1—C3	0.0
N1—C4—C5—C6	-163.23 (11)	C5—C4—N1—C3	180.0
C4 <sup>i</sup> —C4—C5—C6	16.77 (11)	C3 <sup>i</sup> —C3—N1—O1	180.0
C4—C5—C6—C6 <sup>ii</sup>	-96.23 (6)	C2—C3—N1—O1	0.0
C4—C5—C6—C6 <sup>iii</sup>	-17.19 (11)	C3 <sup>i</sup> —C3—N1—C4	0.0
C6 <sup>ii</sup> —C5—C6—C6 <sup>iii</sup>	79.04 (8)	C2—C3—N1—C4	180.0

Symmetry codes: (i)  $-x+1, y, -z+1/2$ ; (ii)  $x, y, -z+1/2$ ; (iii)  $-x+1, y, z$ .

Fig. 1

